Synthesis of 1,4-Thiazepine Derivatives Having  $\text{Angiotensin-Converting Enzyme Inhibitory Property}^{\ 1)}$ 

Hiroaki YANAGISAWA,\* Sadao ISHIHARA, Akiko ANDO, and Takuro KANAZAKI Chemical Research Laboratories, Sankyo Co., Ltd., Hiromachi, Shinagawa-ku, Tokyo 140

Reduction of (6R)-6-amino-6,7-dihydro-2-thienyl-1,4-thiazepin-5(4H)-one derivatives with Mg-MeOH gave the corresponding perhydrothiazepinones. Protection with trityl group at the 6-amino group gave predominantly the 2S,6R isomer which was converted to the potent angiotensin-converting enzyme (ACE) inhibitor. The ACE inhibitor having the 6,7-dihydro-1,4-thiazepin-5(4H)-one ring was also prepared.

Inhibition of angiotensin-converting enzyme (ACE) which plays an important role in renin-angiotensin-aldosterone system results in a lowering of blood pressure and the ACE inhibitors have become antihypertensive agents.  $^{2}$  We have recently reported that the perhydrothiazepinone  $\underline{1}$  had a potent inhibitory activity against ACE and was a candidate for an antihypertensive agent.  $^{3}$  The key intermediate in the synthesis of  $\underline{1}$  is (2S,6R)-6-amino-2-thienylperhydrothiazepin-5-one  $\underline{2}$  which is separated from by-product  $\underline{3}$  by fractional recrystallization. This paper describes the effective utilization of  $\underline{3}$ , that is, the preparation of  $\underline{1}$  from  $\underline{3}$  via the 6,7-dihydro-1,4-thiazepin-5(4H)-ones obtained easily by oxidation of  $\underline{3}$  and the synthesis of an ACE inhibitor having the 6,7-dihydro-1,4-thiazepine ring.

According to the conformational energy calculation of 6-amino-5-oxo-2-phenylperhydrothiazepin-4-ylacetic acid<sup>3)</sup> the 2S,6R isomer  $\underline{6}$  takes a chair conformation possessing two pseudoequatorial substituents at the 2 and 6 positions, while the 2R,6R one  $\underline{4}$  takes the same conformation except for the 6-amino group which is located in a pseudoaxial orientation. Hence it was expected that  $\underline{6}$  would be more predominantly produced than  $\underline{4}$  by reduction of the 6,7-dihydro-1,4-thiazepin-5(4H)-one  $\underline{5}$  which would be derived from  $\underline{4}$  by oxidation.  $\underline{4}$ 

$$\begin{array}{c} R > N \longrightarrow \begin{array}{c} S \longrightarrow S \\ O \longrightarrow CO_2R \end{array} \xrightarrow{ \begin{array}{c} (O) \\ O \longrightarrow CO_2R \end{array}} \begin{array}{c} R > N \longrightarrow \begin{array}{c} S \longrightarrow S \\ O \longrightarrow CO_2R \end{array} \xrightarrow{ \begin{array}{c} (R) \\ O \longrightarrow CO_2R \end{array}} \begin{array}{c} R > N \longrightarrow \begin{array}{c} S \longrightarrow S \\ O \longrightarrow CO_2R \end{array}$$

Treatment of 3 with trityl chloride or di-tert-butyl dicarbonate followed by alkylation with tert-butyl bromoacetate gave 8a,b,<sup>5)</sup> respectively, which were oxidized with tert-butyl hypochlorite in the presence of a base to give 6,7-dihydro-1,4-thiazepin-5(4H)-ones <u>9a,b</u>,<sup>5)</sup> respectively, in moderate yields. was also obtained from 8b through the sulfoxide 10.5 H NMR spectra of 9a,bshow singlets at 6.38 and 6.82, respectively, due to the olefinic proton at the 3 position of the thiazepine ring. Reduction of 9a,b was carried out with magnesium in methanol to give a diastereoisomeric mixture of 8a,b and 11a,bwhich were separated by flash chromatography. The proportion of these isomers is dependent on the protecting group at the 6-amino group, i.e., the trityl derivative 9a gave 8a and 11a in a ratio of 1:4.4, while the tert-butoxycarbonyl (Boc) derivative 9b afforded 8b and 11b in a ratio of 1:1.6. These results indicate that the reduction of the 6,7-dihydro-1,4-thiazepin-5(4H)-one having bulky substituents at the 2 and 6 positions produces preferentially the corresponding perhydrothiazepinone having the pseudoequatorial substituents at the 2 and 6 positions in the chair conformation. Removal of the trityl and Boc groups in 11a,b with aqueous acetic acid and p-toluenesulfonic acid, respectively, gave the amino compound  $12,^{5}$  which was converted to the diester  $14^{3}$ ) by N-alkylation with the sulfonate 13. Removal of the tert-butyl group with acid gave the potent ACE inhibitor 1.3)

Next, the ACE inhibitor  $\underline{17}^{5}$  having the 6,7-dihydro-1,4-thiazepine ring was synthesized from  $\underline{9a,b}$  by the same procedure as in the preparation of  $\underline{1}$  from  $\underline{11a,b}$ , i.e., removal of the amino-protecting group followed by N-alkylation with  $\underline{13}$  and removal of the tert-butyl group. The low yield in  $\underline{17}$  was caused by instability of  $\underline{17}$  in acidic media. Alkaline hydrolysis of  $\underline{17}$  afforded the

dicarboxylic acid  $\underline{18}$  which was less potent than the corresponding dicarboxylic acid  $(IC_{50}=3.6 \text{ nM})^7)$  of  $\underline{1}$  but still had potent inhibitory activity  $(IC_{50}=6.8 \text{ nM}).^7)$ 

$$\frac{3}{2) \text{ iii}} \xrightarrow{\text{R}^1 \text{NH}} \xrightarrow{\text{S}} \xrightarrow{\text{S}} \xrightarrow{\text{iv}} \xrightarrow{\text{R}^1 \text{NH}} \xrightarrow{\text{S}} \xrightarrow{\text{S}} \xrightarrow{\text{Viii}} \xrightarrow{\text{R}^1 \text{NH}} \xrightarrow{\text{N}} \xrightarrow{\text{N}} \xrightarrow{\text{CO}_2 \text{Bu}^t} \xrightarrow{\text{Viii}} \xrightarrow{\text{R}^1 \text{NH}} \xrightarrow{\text{N}} \xrightarrow{\text{N}}$$

a,  $R^1 = Ph_3C$ ; b,  $R^1 = t - BuOCO$  (Boc)

Reagents, reaction conditions and yields. i)  $Ph_{\exists}CC1$ ,  $Et_{\exists}N$  in DMF; r.t., 1.5 h, 84%. ii)  $Boc_{z}0$ ,  $Et_{\exists}N$  in DMF; r.t., 3 h; 85%. iii)  $BrCH_{z}Co_{z}Bu^{\dagger}$ , NaH in DMF; 5 °C, 1 h  $\rightarrow$  r.t., 1 h; 98%. iv) N-Methylmorpholine(1.1 mol equiv.),  $Bu^{\dagger}OC1(1.03 \text{ mol equiv.})$  in toluene; -40--50 °C, 3 h  $\rightarrow$  r.t., 2 h;  $\underline{9a}$ , 81%;  $\underline{9b}$ , 72%. v) mcpba in  $CH_{z}Cl_{z}$ ; -30 °C, 1 h; quantitative. vi)  $Ac_{z}0$ ; 130 °C, 8 h; 52%. Vii) Mg turnings(12 atom equiv.) in MeOH; 30 °C, 4 h, portionwise addition;  $\underline{11a}$ , 65%;  $\underline{8a}$ , 15%;  $\underline{11b}$ , 51%;  $\underline{8b}$ , 30%. viii) For  $\underline{11a}$ ; 80% aq. AcOH; 50 °C, 0.5 h; 83%: for  $\underline{11b}$ ; p-TSA. $H_{z}O$  in dioxane; r.t., 6 h; 96%. ix)  $\underline{13}(1.1 \text{ mol equiv.})$ ,  $Et_{\exists}N(2 \text{ mol equiv.})$  in DMA; 50 °C, 24 h; 96%; x) 4 mol dm<sup>-3</sup> HCl-dioxane; r.t., 16 h; 78%.

Reagents, reaction conditions and yields. i) For  $\underline{9a}$ ; 80% aq. AcOH; 50 °C, 0.5 h; 83%: for  $\underline{9b}$ ; p-TSA.H<sub>2</sub>O in dioxane; r.t., 6 h; 99%. ii)  $\underline{13}$ (1.1 mol equiv.), Et<sub>3</sub>N(2 mol equiv.) in DMA; 50 °C, 24 h; 98%. iii) CF<sub>3</sub>CO<sub>2</sub>H, r.t., 3.5 h followed by neutralization with aq. NaHCO<sub>3</sub>; 15%. iv) 1 mol dm<sup>-3</sup> NaOH; r.t., 5 h; 83%.

## References

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- 3) H. Yanagisawa, S. Ishihara, A. Ando, T. Kanazaki, S. Miyamoto, H. Koike, Y. Iijima, K. Oizumi, Y. Matsushita, and T. Hata, J. Med. Chem., <u>30</u>, 1984(1987).
- 4) Attempts at inversion at the 2 position of the sulfoxide or sulfone of  $\underline{\mathbf{4}}$  through the carbanion at this position were unsuccessful.
- 5) All new compounds were characterized by elemental analyses and spectral dada. The physical and spectral data are as follows. Optical rortations were measured at a 1% solution in DMF at 25 °C. **7a**: amorphous; [  $\alpha$  ]<sub>D</sub>-46.9 °. **7b**: mp 197-199 °C (dec); [  $\alpha$  ] D-76.5 °. **8a**: mp 150-151 °C (dec); [  $\alpha$  ] D-53.2 °. **8b**: mp 162-164 °C ; [  $\alpha$  ] D+3.7 ° .  $\underline{9a}$ : mp 90-91 °C ; [  $\alpha$  ] D-326.8 ° ; NMR (CDC13)  $\delta$ 1.39(9H,s), 3.50(2H,ABq,  $\Delta$   $\delta$  =0.76 ppm,J=17), 3.3-4.0(4H,m), 6.38(1H,s), 6.7-7.7(18H,m). <u>9b</u>: amorphous; [ $\alpha$ ]<sub>D</sub>-603.5°; NMR (CDC1<sub>3</sub>)  $\delta$  1.43(9H,s), 1.46 (3H,s), 3.13(1H,t,J=11), 3.82(1H,d,d,J=5.5,11),  $4.17(2H,ABq,\Delta \delta=0.30 \text{ ppm}$ , J=18), 4.88(1H,d,d,d,J=5.5, 8, 11), 5.84(1H,d,J=8), 6.82(1H,s), 6.9-7.4(3H,s)m). <u>10</u>: mp 201-202 °C; [  $\alpha$  ]<sub>D</sub>-125.6 °. <u>11a</u>: amorphous; [  $\alpha$  ]<sub>D</sub> +28.6 °; NMR (CDCl<sub>3</sub>)  $\delta$  1.45(9H,s), 2.4-3.0(2H,m), 3.70(2H,ABq,  $\Delta$   $\delta$  =0.48 ppm, J=17), 3.55-4.15(3H,m), 4.57(1H,br d,J=7), 6.83(2H,d,J=4), 7.0-7.6(16H,m), 11b: gum; [  $\alpha$  ]<sub>D</sub>+46.3°; NMR (CDCl<sub>3</sub>)  $\delta$  1.47(18H,s), 2.9-3.1(2H,m), 3.5-4.75(3H,m), 4.85-5.15(1H,m),  $6.06(1H,br\ d,J=6)$ , 6.85-7.35(3H,m). <u>12</u>: mp  $81-82\ ^{\circ}C$ ; NMR (CDC1<sub>3</sub>)  $\delta$  1.47(9H,s), 2.29(2H,br s), 2.6-4.7(8H,m), 6.75(3H,m). <u>15</u>: mp 90 °C; [  $\alpha$  ] D-517.5°; NMR (CDC13)  $\delta$  1.47(9H,s), 1.90(2H,br s), 2.78(1H,d,d,J=11, 21), 2.5-4.2(3H,m), 4.17(2H,ABq,  $\Delta$   $\delta$  =0.50 ppm,J=17), 6.89(1H,s), 6.8-7.45 (3H,m). <u>16</u>: syrup; [  $\alpha$  ]<sub>D</sub>-296.8°; NMR (CDC1<sub>3</sub>)  $\delta$  1.20(3H,t,J=7.5),1.47(9H, s), 1.7-2.2(2H,m), 2.3-4.5(11H,m), 6.88(1H,s), 6.8-7.4(3H,m), 7.21(5H,s). 17:mp 146-148 °C; [  $\alpha$  ] D-506.1 °; NMR (CDCl3)  $\delta$  1.16(3H,t,J=7.5), 1.6-2.2(2H, m), 2.3-4.6(10H,m), 6.84(1H,s), 6.7-7.4(3H,m), 7.13(5H,s).  $\underline{18}$ : mp  $215-217\,^{\circ}\text{C}$ (dec); NMR (DMSO-d<sub>6</sub>)  $\delta$  1.6-2.0(2H,m), 2.4-2.8(3H,m), 3.0-3.3(2H,m), 3.55-4.0 (3H,m),  $4.23(2H,ABq, \Delta \delta = 0.32 \text{ ppm}, J=17)$ , 7.0-7.6(9H,m).
- 6) Compounds  $\underline{1}$  and  $\underline{17}$  are the prodrugs which are converted to the active dicarboxylic acids in a living body by enzyme.
- 7) The concentration required for 50% inhibition of rabbit lung ACE with hippurylhistidylleucine as substrate.

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